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### (54) Easy to manufacture meth (acrylic) adhesive compositions

(57) Described are adhesive compositions comprising

said catalyst is present separately, in a further component or paste, respectively,

whereby the amounts of A, B, and C preferably are as follows, namely

B referred to the total amount of A+B = 15 - 60 % by weight, and

C referred to the total amount of A+B+C = 5 - 30 % by weight.

- (i) at least one reactive monomer (A) selected from ethylenically unsaturated carboxylic acid derivatives and mixtures of such derivatives; said one or more derivatives preferably being one or more esters,
- (ii) at least one liquid elastomer in a molar weight range of 1000 - 9000 which is functionalized with ethylenically unsaturated groups (B),
- (iii) at least one impact modifier (C), and
- (iv) at least one free radical initiator and at least one catalyst, whereby optionally either said initiator, or

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## Description

[0001] The present Invention relates to easy to manufacture adhesive compositions based on meth(acrylic) systems comprising a liquid elastomer and an impact modifier.

5 [0002] Methacrylic adhesives combine a number of advantageous properties that make them the ideal solution for many bonding applications. Compared to adhesives based on other chemistries like polyurethanes or epoxies they allow for a relatively long open time with a subsequent fast build-up of strength. Two component methacrylic adhesives reach handling strength within minutes after mixing, and over 80% of the final strength is usually reached within less than an hour. Furthermore, they usually show an excellent adhesion on a wide range of substrates from metals to 10 plastics, like ABS, polyesters, etc. often without pre-treatment. One major requirement an adhesive system needs to fulfill is the durability of the bond. In many applications the long-term performance is significantly increased by the flexibility and the impact resistance of the adhesive. While the two-component meth(acrylic) adhesives of the first generation, formulated with methyl methacrylate, inert fillers e.g. calcium carbonate, catalysts and peroxide initiators show brittle behavior, many efforts have been made in the past to improve flexibility and impact resistance of these 15 systems.

15 [0003] One of said efforts to improve flexibility and impact resistance led to the addition of solid non-reactive elastomers that are dissolved in the meth(acrylic) monomers. Such compositions are called rubber-toughened adhesive compositions. The elastomers are solids at room temperature, and commercially available as large particles or granulates.

20 [0004] This approach is described in several patent documents, such as: US 3,890,407, US 4,106,971, US 4,263,419, US 3,725,504, US 4,200,480, US 3,994,764, EP 0 641 846. The disadvantage of such formulations is that they can only be mixed homogeneously after extremely long dissolving times, or with special equipment like extruders or kneaders, due to the high viscosity of the polymer. Moreover, this type of compositions suffers from the limitation that the monomer, or the monomer mixture, must be chosen such that the non-reactive thermoplastic polymer is soluble therein.

25 Practically, methyl methacrylate is the only monomer with high dissolving properties, and thus the only monomer that allows a polymer-in-monomer composition with up to 30% polymer content. Such low molar mass monomers, like methyl methacrylate, have a strong odor and are highly flammable. Polymer-in-monomer compositions resulting from this approach have a rubbery, stringy consistency and a high viscosity that makes their handling difficult. Due to the high viscosity of the mixture, only a limited amount of fillers can be used making the formulation expensive. A high 30 viscosity is also limiting the adhesion of the formulation, as it limits the wetting of the substrate. Finally, the impact resistance, especially at low temperatures, is low.

35 [0005] Further known are adhesives with multi phase morphology (with core-shell polymers). It is known that composites with a dispersed rubbery phase with a low glass transition temperature have a superior impact resistance. The multi phase morphology can be obtained with core-shell polymers. Core-shell polymers consist of cross-linked acrylic based or butadiene based elastomeric core phase polymers which are grafted onto an outermost rigid thermoplastic polymer. Such core-shell polymers are primarily used as impact modifiers in certain polymer systems like PVC. Because of the cross-linked core such polymers do not dissolve but only swell in the monomers. Such core-shell polymers are e.g. described in US 3,985,703 and US 6,985,704, and adhesive composition with multi phase morphology using core-shell polymers are described in EP 0 087 304.

40 [0006] This approach, however suffers from the limitations that, although these compositions have a high impact-resistance, their elongation at break is low.

[0007] Another known approach is the addition of liquid, low- molar mass elastomers that mix easily with the monomers to form homogeneous liquids. A number of patent documents describe such approach to increase the flexibility of the systems, namely US 4,769,419; US 4,331,765; EP 0,561,352 disclosing mixtures of monomer and liquid rubber.

45 [0008] DE 2,610,423; US 4,439,600; EP 0,640,672; DE 2,319,637; US 4,223,115 disclose mixtures of monomer and acrylic functionalised polyurethane. In US 4,223,115 the formulation optionally contains a dissolved solid elastomer, e.g. NBR (nitrile-butadiene-rubber), polychloroprene.

[0009] The limitation of this approach is that all these compositions lack impact resistance, especially at low temperatures.

50 [0010] Also already known are combinations of solid elastomers, core-shell polymers and monomers. Combinations of solid elastomers and core-shell polymers and monomers are described in US 4,942,201 and US 5,206,288. The dissolved elastomer provides the adhesive with flexibility and the dispersed core-shell graft co-polymer insures a high impact resistance also at low temperatures. The ratios of the three components found in the state of the art are

monomers: solid elastomers: core-shell polymers = A:B:C

55 B in A+B = 5% - 20% or 10% - 35%, and - in another document -

C in A+B+C = 10% - 30%.

[0011] Such compositions show the same limitations as solids in monomers with the exception that they provide improved impact resistance over solid elastomers alone.

[0012] All these systems of the state of the art have limited properties with regard to flexibility and/or impact resistance, or they are difficult to manufacture.

[0013] The goal of the present invention therefore was to provide adhesive compositions with good flexibility and impact resistance that are furthermore easy to manufacture.

5 [0014] It has now surprisingly been found that adhesive compositions with very good flexibility and impact resistance are obtained if they comprise

- (i) at least one reactive monomer (A) selected from ethylenically unsaturated carboxylic acid derivatives and mixtures of such derivatives, said one or more derivatives preferably being one or more esters,
- 10 (ii) at least one liquid elastomer in a molar weight range of 1000 - 9000 which is functionalized with ethylenically unsaturated groups (B),
- (iii) at least one impact modifier (C), and
- (iv) at least one free radical initiator and at least one catalyst, whereby optionally either said initiator, or said catalyst is present separately, in a further component or paste, respectively.

15 [0015] Optionally the composition can further comprise at least one adhesion promotor like (meth)acrylic acids, (meth)acrylic phosphate esters, silanes such as 3-mercaptopropyltrimethoxysilane, (3-methacryloyloxypropyl)trimethoxysilane and 3-glycidyloxypropyltrimethoxysilane or the respective triethoxysilanes, or mixtures of said silanes, and/or Zn di(meth)acrylates.

20 [0016] In a much preferred embodiment, the adhesive composition of the present invention comprises the components A, B, and C in the following amounts:

B based on the total weight of A+B in an amount of 15 - 60 % by weight, preferably 15 - 40 % by weight, and

[0017] C based on the total weight of A+B+C in an amount of 5 to 30 % by weight, preferably 10 to 25 % by weight.

25 [0018] The reactive monomers (A), preferably are esters. Preferred esters are (meth)acrylic esters, in particular linear or branched or cyclic C<sub>1</sub>-C<sub>6</sub>-alkyl esters or heterocyclic or aromatic esters. Much preferred esters are methyl-methacrylate (MMA), tetrahydrofurfurylmethacrylate (THFMA), cyclohexylmethacrylate (CHMA), and cyclic trimethylolpropane formal acrylate (CTFA), isobornylmethacrylate (IBMA), benzylmethacrylate (BMA), dicyclopentadienyl-oxoethylmethacrylate (DCPOEMA), t-butylmethacrylate (tBMA), isobornylacrylate (IBH), dihydrodicyclopentadienylacrylate (DHDCPA).

30 [0019] Examples for free radical initiators are organic peroxides, in particular benzoylperoxide, and examples for catalysts are tertiary amines and/or salts and/or complexes of transition metals. Examples for tertiary amines are N, N-dimethylaniline, N,N-dimethyl-p-toluidine, N,N-diethylaniline, N,N-diethyltoluidine, N,N-bis(2-hydroxyethyl)-p-toluidine, N-ethoxylated p-toluidine, N-alkylmorpholine or mixtures thereof, and examples for the salts and complexes of transition metals are salts and complexes of cobalt, nickel and/or copper.

35 [0020] Preferred adhesive composition of the present invention are compositions wherein the reactive monomer comprises at least 50 % of (meth)acrylic monomer, and whereby at least 50% of said (meth)acrylic monomer is tetrahydrofurfuryl methacrylate or methyl methacrylate or mixture thereof. In a much preferred embodiment, at least 50 % of the reactive monomer(s) is tetrahydrofurfuryl methacrylate or methyl methacrylate or mixture thereof.

40 [0021] It is also preferred that the homo- or copolymers resulting from said reactive monomers A have a glass transition temperature of from about 40°C to about 140°C.

[0022] With regard to the liquid elastomer (B), said elastomer is preferably chosen from the group consisting of

- (meth)acrylic functionalized butadiene, isoprene based polymers or block-copolymers, and
- PU-(meth)acrylate obtainable through the syntheses of a polyethylene polyol or polypropylene polyol, a diisocyanate and a hydroxy functionalized ethylenically unsaturated monomer, and
- mixtures thereof.

45 [0023] A preferred PU (meth)acrylate is obtainable through a synthesis using polyols with low unsaturation and narrow molar weight distribution as obtainable through double metal cyanide complex catalysis (Acclaim polyols). As well preferred as liquid elastomers are (meth)acrylic terminated butadiene acrylonitrile block copolymers.

[0024] The impact modifier (C) is usually a core shell polymer which does not dissolve, but swells in the monomer. suitable core-shell polymers contain a cross-linked acrylic or butadiene based elastomeric core phase polymer which is grafted onto a rigid thermoplastic shell phase polymer.

55 [0025] The compositions of the present invention can comprise further components well known to the skilled person such as preferably also at least one organic or inorganic filler or thixotropic agent. The compositions can also comprise further substances, such as stabilizers, additives, toughening agents, adhesion promoters, defoaming agents, thickeners, plasticizers, wetting agents, wax compounds, cross-linking agents, inhibitors etc. Such additional substances are known to the skilled person. Examples for inhibitors are e.g. hydrochinone, methylhydrochinone, t-butyl-p-cresol

and for thixotropic agents e.g. Aerosil.

[0026] The compositions of the present invention can be easily manufactured in a short time. To optimize the consistency of the paste it has proved to be advantageous to usually start with only part, usually about half of the amount of reactive monomer A and to add the second part at a later stage.

5 [0027] A usual procedure is as follows: A first part (or if viscosity allows it all) of reactive monomer A, the liquid elastomer B and catalyst and/or initiator D are mixed under heating and stirring until a homogenous mass is obtained (at a temperature of about 60°C for about 30 minutes). Then the whole amount of impact modifier C is added at once, and the mixture is stirred for approx. another 30 minutes. Finally, the second part of reactive monomer A is added and the composition is stirred for another about 30 minutes under vacuum.

10 [0028] Usually either the catalyst or the initiator D are provided in a separate component unless at least one of them is present in a non-reactive form so that it has first to be activated, e.g. by heating.

[0029] The invention is now further described by means of examples. Said examples, however, shall not limit the scope of the present invention in any way.

15 **Examples**

**Preparation of compositions**

[0030] The components used in the examples are identified as:

20	MMA	Methyl methacrylate
	THFMA	Tetrahydrofurfuryl methacrylate
25	CHMA	Cyclohexyl methacrylate
	CN 965	aliphatic polyurethane-acrylate with a MW 5600 from Cray Valley
	Hycar VTBNX 1300 X33	acrylic functionalized butadiene-acrylo nitrile blockcopolymer from BFGoodrich
30	Paraloid EXL 2600	core shell polymer from Rohm and Haas
	p-Toluidine	N,N-bis(2-hydroxyethyl)-p-toluidine
35	BPO Paste	a paste of 40 % benzoyl peroxide in phthalate plasticizer

[0031] The components used to formulate the adhesive compositions of Examples 1 to 7 and their amounts are listed in Table 1.

40 **The composition was produced as follows:**

[0032] Half the amount of reactive monomer A, the liquid elastomer B and catalyst D are added to a vessel equipped with a stirrer, a heating device and a vacuum device. The heating is turned on, and the mixture is stirred for about 30 minutes at 60°C until a homogenous mass was obtained. Then, the heating is turned off, the whole amount of the impact modifier C is added at once, and the mixture is stirred for approx. 30 minutes. Finally, the remaining amount of reactive monomer A is added and the composition stirred for about 30 minutes under vacuum.

**For 1 kg of the composition of e.g. Example 2 the above described procedure was applied as follows:**

50 [0033] 265 g of tetrahydrofurfuryl methacrylate are added together with 260 g of Hycar and 10 g of p-toluidine into a 1 kg lab dissolver equipped with an anchor stirrer and a heating and vacuum device. The heating is turned on and the mixture is stirred for about 30 minutes at 60°C until a homogenous mass is obtained. Then the heating is turned off, 200 g of Paraloid EXL 2600 are added at once, and the mixture is stirred for approx. 30 minutes. Finally, another 265 g of tetrahydrofurfuryl methacrylate are added and stirred for about 30 minutes under vacuum.

55 [0034] The adhesives were cured by adding 4 % of the BPO paste.

[0035] The test results obtained for the cured adhesives are also listed in Table 1.

The adhesives were tested as follows:

**Test procedures:**

5 **Tensile strength and Elongation (DIN 53504)**

[0036] The adhesive was cured to form a sheet of approximately 2.5 mm thickness from which tensile test dumbbells were cut. The stress-strain tests were performed using a rate of 200 mm per minute for measurements at room temperature.

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**Impact Test or Charpy Test (DIN 53453)**

[0037] The impact strength of the adhesive was tested using adhesive bars of 11.8 cm length, 12 cm width and 9.5 cm height. The impact strength was measured at the indicated temperature with a pendulum of 15 kg

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**Table 1**

	<i>Ex 1</i>	<i>Ex 2</i>	<i>Ex 3</i>	<i>Ex 4</i>	<i>Ex 5</i>	<i>Ex 6</i>	<i>Ex 7</i>	<i>Designation</i>
MMA	57			43	70		70	A
THFMA		53	18			80		
CHMA			41					
EHA	0		7					
CN 965	18			43	7			C
Hycar VTNB 1300 X33		26	20			19		
Paraloid EXL 2600	23	20	14	13	22		29	
p-Toluidine	1	1	1	1	1	1	1	D
%B/A+B	24	33	17	50	9	19	0	
TS [Mpa]	18	9	15	11	34	15	21	
EaB [%]	135	210	155	219	19	140	28	
Impact at -20°C [mmJ/mm <sup>2</sup> ]	35	40	25	30	25	< 1	< 5	

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[0038] Examples 1 to 4 are compositions which provide adhesives with a high elongation and resistance to low temperatures. Example 5 is made by using a too low amount of liquid elastomer which results in an adhesive with a good low temperature impact resistance, but with a too low elongation. In example 6, a formulation is represented which does not contain any core-shell polymer. Although the elongation is high, the impact resistance of the material at low temperatures is low. Example 7, which is a composition containing only core shell polymer and no liquid elastomer, has not only a low elongation but also a reduced impact resistance. As this formulation does not contain any liquid elastomer (compared with example 5 which contains a low amount of elastomer), the interface between polymer matrix and the impact modifier is not coherent enough to transfer the energy from the matrix onto the impact modifier.

## Claims

1. An adhesive compositions comprising
  - 5 (i) at least one reactive monomer (A) selected from ethylenically unsaturated carboxylic acid derivatives and mixtures of such derivatives,
  - (ii) at least one liquid elastomer in a molar weight range of 1000 - 9000 which is functionalized with ethylenically unsaturated groups (B),
  - (iii) at least one impact modifier (C), and
  - 10 (iv) at least one free radical initiator and at least one catalyst, whereby optionally either said initiator ,or said catalyst is present separately, in a further component or paste, respectively.
2. The adhesive composition of claim 1 wherein the amount of B based on the total amount of A+B is 15 - 60 % by weight, preferably 15 - 40 % by weight.
- 15 3. The adhesive composition of claim 1 or 2 wherein the amount of C based on the total amount of A+B+C is 5 - 30 % by weight, preferably 10 - 25 % by weight.
4. The adhesive composition of anyone of the preceding claims, wherein said at least one ethylenically unsaturated carboxylic acid derivative is an ester or a mixture of esters.
- 20 5. The adhesive composition of anyone of the preceding claims, wherein the reactive monomer is a (meth)acrylic monomer.
6. The adhesive composition of anyone of the preceding claims, wherein the reactive monomer comprises at least 50% of tetrahydrofurfuryl methacrylate or methyl methacrylate or mixtures thereof.
- 25 7. The adhesive composition of anyone of the preceding claims, wherein the glass transition temperature of the homo- or copolymer resulting from said reactive monomers A is between 40°C and 140°C.
- 30 8. The adhesive composition of anyone of the preceding claims, wherein the liquid elastomer (B) is chosen from the group consisting of
  - (meth)acrylic functionalized butadiene, isoprene based polymers or block-copolymers, and
  - PU-(meth)acrylate obtainable through the syntheses of a polyethylene polyol or polypropylene polyol, a diisocyanate and a hydroxy functionalized ethylenically unsaturated monomer, and
  - mixtures thereof.
- 35 9. The adhesive composition of claim 8, wherein the PU (meth)acrylate is obtainable through a synthesis using polyols with low unsaturation and narrow molar weight distribution as obtainable through double metal cyanide complex catalysis.
- 40 10. The adhesive composition of claim 8 or 9, wherein the liquid elastomer is a (meth)acrylic terminated butadiene acrylonitrile blockcopolymer.
- 45 11. The adhesive composition of anyone of the preceding claims, wherein the impact modifier (C) is a core shell polymer which does not dissolve but swells in the monomer.
12. The adhesive composition of anyone of the preceding claims wherein the initiators is an organic peroxide, in particular benzoylperoxide, and the catalyst is selected from the group consisting of tertiary amines, and salts of transition metals and complexes of transition metals, in particular the catalyst is selected from the group consisting of N,N-dimethylaniline, N,N-dimethyl-p-toluidine, N,N-diethylaniline, N,N-diethyltoluidine, N,N-bis(2-hydroxyethyl)-p-toluidine, N-ethoxylated p-toluidine, N-alkylmorpholine or mixtures thereof, or from the group consisting of salts and complexes of cobalt, nickel and copper and mixtures thereof.
- 50 13. The adhesive composition of anyone of the preceding claims which contains an adhesion improving agent, in particular an adhesion impeding agent selected from the group consisting of (meth)acrylic acids, (meth)acrylic phosphate esters, silanes such as 3-mercaptopropyltrimethoxysilane, (3-methacryloyloxypropyl)trimethoxysilane

and 3-glycidyloxypropyltrimethoxysilane or the respective triethoxysilanes, or mixtures of said silanes, and/or Zn di(meth)acrylates.

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## EUROPEAN SEARCH REPORT

Application Number  
EP 01 10 5349

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.7)
X	US 4 439 600 A (MORAN JR JAMES P) 27 March 1984 (1984-03-27) * column 11, line 38-42 *	1-5, 7-9, 12, 13	C09J4/06 C09J175/16 C08F290/06
X	WO 00 77109 A (LORD CORP) 21 December 2000 (2000-12-21) * examples *	1-8, 12, 13	
X	EP 1 013 735 A (THREE BOND CO LTD) 28 June 2000 (2000-06-28) * claim 1; examples *	1-9, 11-13	
X	WO 00 73397 A (LORD CORP) 7 December 2000 (2000-12-07) * the whole document *	1-5, 8, 10, 12, 13	
A	US 5 710 235 A (ABBEY KIRK JOSEPH ET AL) 20 January 1998 (1998-01-20) * column 8, line 51-54; claims 1, 2, 11 *	1	
A	WO 95 25760 A (DSM NV) 28 September 1995 (1995-09-28) * claim 1 *	1	TECHNICAL FIELDS SEARCHED (Int.Cl.7) C09J C08F
<p>The present search report has been drawn up for all claims</p>			
Place of search	Date of completion of the search	Examiner	
THE HAGUE	31 July 2001	Meulemans, R	
CATEGORY OF CITED DOCUMENTS		T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document	
X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document			

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ON EUROPEAN PATENT APPLICATION NO.

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This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report. The members are as contained in the European Patent Office EDP file on The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

31-07-2001

Patent document cited in search report		Publication date		Patent family member(s)	Publication date
US 4439600	A	27-03-1984		AU 2838484 A CA 1216982 A EP 0130680 A NO 842212 A ZA 8403797 A	06-12-1984 20-01-1987 09-01-1985 04-12-1984 24-12-1984
WO 0077109	A	21-12-2000		AU 5327300 A	02-01-2001
EP 1013735	A	28-06-2000		JP 10237420 A BR 9812422 A US 6232431 B CN 1248281 T WO 9837161 A	08-09-1998 26-09-2000 15-05-2001 22-03-2000 27-08-1998
WO 0073397	A	07-12-2000		US 6225408 B AU 5037800 A	01-05-2001 18-12-2000
US 5710235	A	20-01-1998		AU 1129697 A CN 1202908 A EP 0866809 A JP 2000502136 T WO 9721735 A	03-07-1997 23-12-1998 30-09-1998 22-02-2000 19-06-1997
WO 9525760	A	28-09-1995		AU 1962995 A US 6021338 A	09-10-1995 01-02-2000